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Within the actual stage we have continued with (i) the study of up-conversion effects shown by $(Yb^{3+}-Er^{3+})$ doped SiO₂-YLiF₄ glass-ceramics nanorods and (ii) we have prepared Ho³⁺-doped SiO₂-YLiF₄ glass-ceramics nanorods.



Figure 1. Green $({}^{2}H_{11/2}, {}^{4}S_{3/2} \rightarrow {}^{4}I_{15/2})$ and red $({}^{4}F_{9/2} \rightarrow {}^{4}I_{15/2})$ Er³⁺ up-conversion luminescences excited at 980 nm in Yb³⁺-Er³⁺-doped oxyfluoride glass-ceramic "bulk" or nanorods and LiYF₄ pellet samples.

Up-conversion properties of Yb³⁺/Er³⁺ co-doped LiYF₄ nanocrystals in glass-ceramics "bulk" or nanorods have been studied by comparison to the corresponding glass-ceramic. Under 980 nm laser light pumping the green emitting levels (${}^{2}H_{11/2}$, ${}^{4}S_{3/2}$) are populated by the Yb³⁺ - Er³⁺ ET1 [${}^{2}F_{5/2}$ (Yb³⁺) + ${}^{4}I_{15/2}$ (Er³⁺) $\rightarrow {}^{2}F_{7/2}$ (Yb³⁺) + ${}^{4}I_{11/2}$ (Er³⁺)] and ET2 [${}^{2}F_{5/2}$ (Yb³⁺) + ${}^{4}I_{11/2}$ (Er³⁺) $\rightarrow {}^{2}F_{7/2}$ (Yb³⁺) + ${}^{4}F_{7/2}$ (Yb³⁺) + ${}^{4}F_{7/2}$ (Er³⁺)] processes (Figures 1-2), followed by the rapid multiphonon (MP) transition ${}^{4}F_{7/2} \rightarrow {}^{2}H_{11/2}$. The lower emitting levels are then populated via multiphonon and cross-relaxation (Yb³⁺-Er³⁺) processes followed by the green ((${}^{2}H_{11/2}$, ${}^{4}S_{3/2}$) $\rightarrow {}^{4}I_{15/2}$) and red (${}^{4}F_{9/2} \rightarrow {}^{4}I_{15/2}$) luminescences.



Figure 2. Energy level schemes of Yb³⁺ and Er³⁺ and the main energy transfer processes from Yb³⁺ to Er³⁺ represented by curved dashed arrows.



Figure 3. Double logarithmic plot of luminescence intensity vs. incident laser power for green (${}^{2}H_{11/2}$, ${}^{4}S_{3/2}$) $\rightarrow {}^{4}I_{15/2}$) and red (${}^{4}F_{9/2} \rightarrow {}^{4}I_{15/2}$) luminescences recorded on Yb³⁺/Er³⁺ co-doped LiYF₄ glass ceramic "bulk" (or nanorods) and LiYF₄ pellet samples.

Laser pump power dependencies of UC luminescence intensities show a quadratic laser power dependence indicating a two-photon process (Figure 3). Photoluminescence lifetimes measurements have shown a drastically reduction of the 540nm "green" luminescence decay time from 0.015±0.001 ms in the glass ceramic bulk to to about 0.001ms in the naorods that indicates the influence of the dimensional constraints imposed by the membrane pores during xerogel formation and glass ceramization.

Preparation of the Ho-doped glass-ceramics nanorods was made by using the template method and nanoporous polycarbonate membranes (pore diameter size 800nm). The method implies the filling of the pores of the membrane and nanorods/nanowires followed solidification and membrane separation by dissolving. RE-doped gels ($95SiO_2-5LiYF_4$) (mol%) doped with rare earths Yb (4mol% mol.) and Ho(1% mol) have been prepared by using tetraethylorthosilicate (TEOS) as precursor and trifluoroacetic acid (TFA) as fluorine agent; Yb and Ho acetates have been used for doping. For the first solution that imply (TEOS-tetraethoxysilane) precursor liquid we used two different molar ratio for the ethanol to water: TEOS:Et:H2O:CH3COOH is 1 : 10 : 4 : 0.5 (A synthesis) si TEOS:Et:H2O:CH3COOH is 1 : 4 : 10 : 0.5 (B synthesis). For the glass ceramization we have used a thermal tratement at higher temperatures. Crystallization mechanism involves the formation of the nucleating centres due to the thermolysis (thermal decomposition) of the Y(CF₃COO)₃ Li(CF₃COO) acetates with the formation of YF₃ and LiF, followed at higher temperatures (530 °C) by the LiYF₄ nanocrystals formation with the Ho si Yb dopants incorporation (Figure 4).

Morphological and structural characterisation of the glass-ceramics nanorods has been made by using Xray difraction (XRD) and scanning electron microscopy (SEM) measurements. The small quantities of the material (glass ceramic rods) and sensitivity of the x-ray diffraction technique (of about 2-3%) did not allow to measure the LiYF₄ crystallized fraction within the rods and therefore these studies have been performed on "bulk" (powder samples) based on the supposition that the crystallization processes are the same within the "bulk" and rods. The XRD pattern of the glass ceramics "bulk" has shown diffraction peaks due to the LiYF₄ crystalline phase precipitation in the glass matrix. From the analysis of the XRD pattern we have extracted the mean size of the nanocrystals which is d \cong 55-60nm (A synthesis) and d \cong 25-30nm (B synthesis). On the other hand the crystallization fraction is about 40% for the A synthesis and only 20% for the B synthesis. SEM images have shown that in the first case the nanorods are partially agglomerated and are curved or bent; in both cases are formed LiYF₄ nanocryslas as the UP-conversion luminescence spectra have shown.



Figure 4. SEM images of the oxyfluoride glass-ceramic SiO2-LiYF4(Yb,RE) microrods.

In conclusion, up-conversion properties of Yb³⁺/Er³⁺ co-doped LiYF₄ nanocrystals are similar in the glassceramics ("bulk" or nanorods) and pellet showing green ((${}^{2}H_{11/2}$, ${}^{4}S_{3/2}$) $\rightarrow {}^{4}I_{15/2}$) and red (${}^{4}F_{9/2} \rightarrow {}^{4}I_{15/2}$) up-conversion luminescence bands due to the Er³⁺ ions. Laser pump power dependencies of UC showed a quadratic laser power dependence indicating a two-photon process. The non-radiatie decay processes are much higher in the glass ceramic rods compared to the bulk an effect that might be related to the quasi-unidimensional morphology of the nanorods.

Regarding the synthesis of the Ho^{3+} doped YLF (YLiF₄) nanorods it was shown that the molar ratio ethanol/water has a great influence on the crystallization fraction but weaker on the nanocrystals size.

References:

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